



**Quarter 1 (45 Days)**

| Chapter  | Topic  | Dates               | # Days                           |
|----------|--|---------------------|----------------------------------|
| <b>1</b> | <b>Nature of Science</b> <ul style="list-style-type: none"> <li>The Methods of Science</li> <li>Standards of Measurement</li> <li>Communicating with Graphs</li> <li>Science and Technology</li> </ul>   | 8-12<br>To<br>9-4   | 18                               |
| <b>2</b> | <b>Motion</b> <ul style="list-style-type: none"> <li>Introduction to One-Dimensional Vectors</li> <li>Displacement and Distance</li> <li>Velocity (constant, average, and instantaneous)</li> <li>Acceleration</li> <li>Interpreting Position vs. Time and Velocity vs. Time Graphs</li> </ul>   | 9-8<br>To<br>9-18   | 8                                |
| <b>3</b> | <b>Forces</b> <ul style="list-style-type: none"> <li>Force Diagrams</li> <li>Types of Forces (gravity, friction, normal, tension)</li> <li>Field Model for Forces at a Distance</li> <li>Newton's 3 Laws of Motion</li> <li>Dynamics (how forces affect motion)</li> <li>Forces on Objects at Rest</li> <li>Forces on Objects Moving with Constant Velocity</li> <li>Forces on Accelerating Objects</li> </ul> | 9-21<br>To<br>10-2  | 9                                |
| <b>4</b> | <b>Work and Energy</b> <ul style="list-style-type: none"> <li>Conservation of Energy</li> <li>Quantifying Kinetic Energy</li> <li>Quantifying Gravitational Potential Energy</li> </ul>  | 10-5<br>To<br>10-13 | 7                                |
| <b>5</b> | <b>Thermal Energy</b> <ul style="list-style-type: none"> <li>Specific Heat</li> <li>Transfer and Transformation of Energy (including work)</li> </ul>  | 10-14-<br>10-20     | 5<br>(overlaps and goes into Q2) |

**Quarter 2 (35 Days, 4 Exam Days)**

|  |  |                      |    |
|--|--|----------------------|----|
| <b>6</b>   | <b>Electricity and Magnetism</b> <ul style="list-style-type: none"> <li>Movement of Electrons</li> <li>Current</li> <li>Electric Potential (voltage)</li> <li>Resistors and Transfer of Energy</li> <li>Electric Circuits</li> <li>Electrical Conductors and Insulators</li> <li>Overview of Magnetism</li> </ul>  | 10-21<br>To<br>11-13 | 16 |
| <b>9</b><br><b>10.1; 10.2</b><br><b>11.1; 11.2</b> | <b>Waves</b> <ul style="list-style-type: none"> <li>Wavelength, Frequency, and Speed of a Wave</li> <li>Wave Reflection</li> <li>Wave Refraction</li> <li>Wave Diffraction</li> <li>Absorption</li> <li>Superposition (constructive and destructive interference)</li> <li>Radiant Energy and the Electromagnetic Spectrum</li> <li>Doppler Shift</li> </ul> | 11-16<br>To<br>12-9  | 15 |
|  | <b>Semester 1 Exam Review and Exam</b> <ul style="list-style-type: none"> <li>Chapters 1-5, 9-10</li> </ul>  | 12-10<br>To<br>12-17 | 6  |

**Quarter 3 (45 Days)**

| Chapter         | Topic   | Dates              | # Days |
|-----------------|---|--------------------|--------|
| <b>14-1; 15</b> | <b>Classification of Matter</b> <ul style="list-style-type: none"> <li>Heterogeneous vs. Homogeneous Mixtures</li> <li>Physical and Chemical Properties of Matter</li> <li>Solutions</li> <li>Phase Changes</li> <li>Endothermic and Exothermic Processes of Phase Changes</li> <li>Calculation of Density from Mass vs. Volume Graphs of Substances</li> </ul> | 1-4<br>To<br>1-26  | 16     |
| <b>16</b>       | <b>Atoms and the Periodic Table</b> <ul style="list-style-type: none"> <li>Models of the Atoms and its Components</li> <li>Atomic Number and Mass Number</li> <li>Ions (cations and anions)</li> <li>Isotopes</li> <li>The Periodic Table and Periodic Law</li> <li>Representative Groups in the Periodic Table</li> </ul>                                      | 1-27<br>To<br>2-18 | 15     |
| <b>18</b>       | <b>Chemical Bonds and Compounds</b> <ul style="list-style-type: none"> <li>Ionic Bonding</li> <li>Covalent Bonding</li> <li>Prediction of Ionic Charge and Formulas of Ionic Compounds using Elements from Groups 1, 2, 17, Hydrogen and Oxygen</li> <li>Naming a Chemical Compound when given a Chemical Formula</li> </ul>                                    | 2-19<br>To<br>3-11 | 16     |

**Quarter 4 (46 Days)**

|           |   |                    |    |
|-----------|---|--------------------|----|
| <b>19</b> | <b>Chemical Reactions</b> <ul style="list-style-type: none"> <li>Conservation of Mass</li> <li>Writing Balanced Chemical Equations</li> <li>Identifying Reactants and Products in a Chemical Reaction</li> <li>Balancing Chemical Equations</li> <li>Endothermic and Exothermic Chemical Reactions</li> </ul> | 3-21<br>To<br>4-8  | 14 |
| <b>20</b> | <b>Nuclear Reactions</b> <ul style="list-style-type: none"> <li>Strong and Weak Nuclear Forces</li> <li>Radioactive Decay</li> <li>Application of Radioactive Isotopes in Medicine</li> <li>Half-life of a radioisotope</li> <li>Nuclear Fissions and Fusion</li> </ul>                                       | 4-11<br>To<br>4-26 | 12 |
| <b>31</b> | <b>Stars and Galaxies</b> <ul style="list-style-type: none"> <li>History of the Universe</li> <li>Galaxy Formation and Classification</li> <li>Star Formation and Evolution</li> <li>Nuclear Fusion in Stars</li> <li>How Stars are Classified</li> </ul>   | 4-27<br>To<br>5-18 | 16 |
|           | <b>Semester 2 Exam Review and Exam (Chapters 14-1, 15, 16, 18, 19, 20, 31)</b>  | 5-19<br>To<br>5-25 | 5  |